

Xue-Mei Li, Shan-Shan Gu and
Shu-Sheng Zhang*College of Chemistry and Molecular Engineering,
Qingdao University of Science and
Technology, 266042 Qingdao, Shandong,
People's Republic of China

Correspondence e-mail: shushzhang@126.com

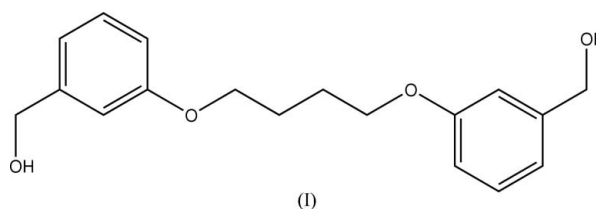
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.052
 wR factor = 0.156
Data-to-parameter ratio = 15.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

1,4-Bis(3-hydroxymethylphenoxy)butane

The molecule of the title compound, $\text{C}_{18}\text{H}_{22}\text{O}_4$, has a center of symmetry. Molecules are linked into a zigzag chain *via* an $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond. The chains are further linked by $\text{O}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions to form a layer.Received 9 January 2006
Accepted 13 January 2006
Online 20 January 2006

Comment

We have recently reported the structure of 1,6-bis(3-hydroxymethylphenoxy)hexane, (II) (Li *et al.*, 2006). In our ongoing studies of diethers, the title compound, (I), was synthesized and its crystal structure was determined.The title molecule has a center of symmetry at the mid-point of the central $\text{C}-\text{C}$ bond. All bond lengths and angles in (I) (Table 1) show normal ranges (Allen *et al.*, 1987), and are comparable to those in (II). In the crystal structure, molecules are linked into a zigzag chain along the $[201]$ direction by a weak $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond (Fig. 2 and Table 2). The chains are further linked by $\text{O}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions (Table 2), forming a layer parallel to the (201) plane.

Experimental

To a solution of 3-hydroxymethylphenol (1.5 g, 12 mmol) in EtOH (90 ml) was added 10 mol l^{-1} NaOH (1.2 ml, 12 mmol), followed by 1,4-dibromobutane (1.3 g, 6 mmol). The reaction was refluxed for 23 h, cooled to room temperature and diluted with water (30 ml). The brown mixture was subsequently extracted with CH_2Cl_2 , and the combined extracts were dried (MgSO_4) and concentrated *in vacuo* to give an off-white solid. Purification by flash chromatography ($\text{CH}_2\text{Cl}_2/\text{Me}_2\text{CO}$ 4:1, *v/v*) gave the title compound as a slightly pink solid. Yellow single crystals were obtained by slow evaporation of an ethyl acetate–ethanol (3:1 *v/v*) solution at room temperature over a period of two weeks.

Crystal data

 $\text{C}_{18}\text{H}_{22}\text{O}_4$
 $M_r = 302.36$
Monoclinic, $P2_1/c$
 $a = 13.920$ (3) Å
 $b = 4.8454$ (9) Å
 $c = 12.846$ (2) Å
 $\beta = 112.347$ (3)°
 $V = 801.4$ (3) Å³
 $Z = 2$ $D_x = 1.253$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 1160
reflections
 $\theta = 3.2\text{--}23.7^\circ$
 $\mu = 0.09$ mm⁻¹
 $T = 293$ (2) K
Plate, yellow
 $0.36 \times 0.16 \times 0.06$ mm

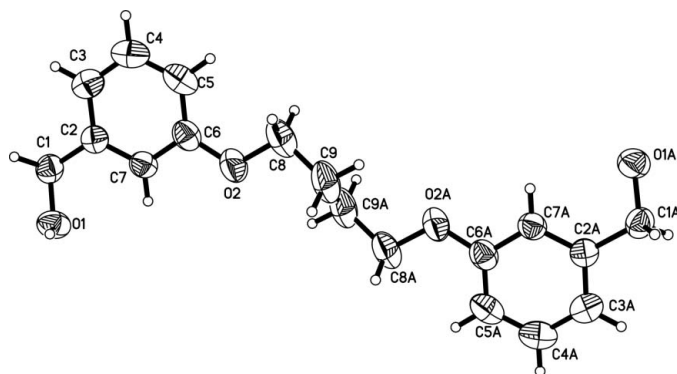


Figure 1
The structure of (I), showing 50% probability displacement ellipsoids and the atom numbering scheme. The suffix A corresponds to symmetry code $(1 - x, 2 - y, 2 - z)$.

Data collection

Siemens SMART 1000 CCD area-detector diffractometer	1557 independent reflections
ω scans	1186 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.020$
$T_{\text{min}} = 0.969$, $T_{\text{max}} = 0.995$	$\theta_{\text{max}} = 26.0^\circ$
4195 measured reflections	$h = -17 \rightarrow 15$
	$k = -5 \rightarrow 5$
	$l = -15 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0758P)^2 + 0.191P]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.156$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.33 \text{ e \AA}^{-3}$
1557 reflections	$\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$
100 parameters	
H-atom parameters constrained	

Table 1

Selected bond lengths (\AA).

O1—C1	1.405 (2)	O2—C8	1.437 (2)
O2—C6	1.367 (3)	C8—C9	1.497 (3)

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$ is the centroid of the C2–C7 benzene ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1A \cdots O1 ⁱ	0.82	1.95	2.765 (2)	173
C1—H1C \cdots Cg1 ⁱⁱ	0.97	2.76	3.685	160

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x, y - 1, z$.

All H atoms were located in a difference Fourier map and constrained to ride on their parent atoms, with $O-H = 0.82 \text{ \AA}$ and $C-H = 0.93-0.97 \text{ \AA}$, and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ and $1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

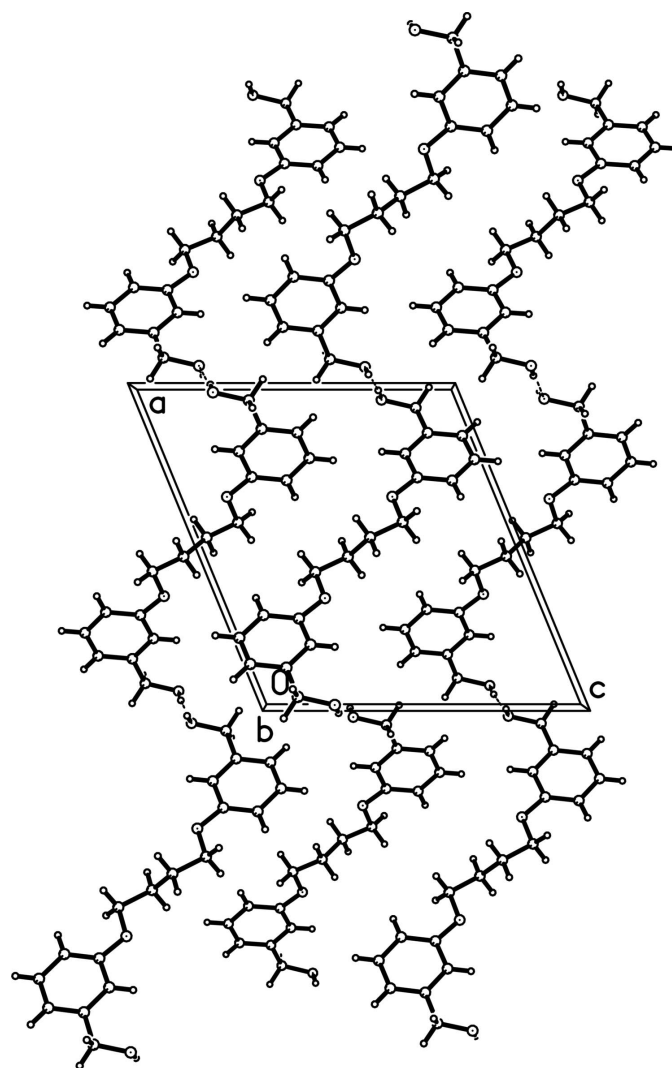


Figure 2

A packing view down the b axis. Hydrogen bonds are indicated by dashed lines.

This project was supported by the Project of Educational Administration of Shandong Province (No. J04B12) and the Outstanding Adult-Young Scientific Research Encouraging Foundation of Shandong Province (No. 2005BS04007).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Li, X.-M., Gu, S.-S. & Zhang, S.-S. (2006). *Acta Cryst. E* **62**, o487–o488.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.